

# Study Guide: Physical Science

## Lab and Lab Safety



- Follow procedures
- Use materials appropriately
- Think

## Scientific Method

Purpose/Question  
Research  
Hypothesis  
Experiment  
Analysis and Conclusion

**Variables**- any factor in an experiment that can be changed  
**Constants**- variables that don't change in the experiment  
**Independent variable**- the variable being changed in the experiment  
**Dependent Variable**- the variable that is being measured  
**Control**- a reference for comparison in an experiment  
**Inference**- a conclusion based on the results of the experiment

## Lab, Mass, Volume, and Density



*A triple beam balance* measures mass in grams

*A graduated cylinder* measures volume in ml

*A graduated cylinder* also measures the volume of solids by liquid displacement



Ice floats in water because its density is less than liquid water



$$\text{Density} = \frac{\text{Mass}}{\text{Volume}}$$

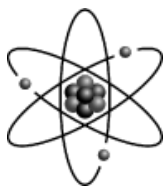
**Mass** is the amount of matter in an object

**Volume** is the amount of space an object occupies

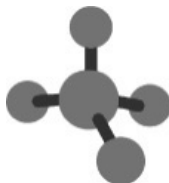
## Atoms, Elements, Compounds

- Matter includes all solids, liquids and gases on Earth. All matter is made of atoms.

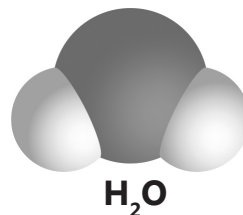
Atom



Compound



Water Molecule



Compounds are made of two or more atoms

## Periodic Table

- A chart that organizes the chemical elements in order of atomic number and their chemical characteristics.

Elements in a column have similar chemical properties

Most elements are metals

**Metals**  
conduct electricity  
good conductors of heat  
Shiny

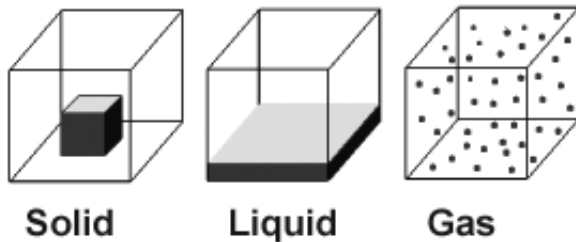
Increasing atomic number in a row

|                               |                               |                                |                                |                                 |                               |                                |                                   |
|-------------------------------|-------------------------------|--------------------------------|--------------------------------|---------------------------------|-------------------------------|--------------------------------|-----------------------------------|
|                               |                               |                                |                                |                                 |                               |                                | 18<br>helium<br>2<br>He<br>4.0026 |
|                               |                               |                                |                                |                                 |                               |                                | 17                                |
|                               |                               | boron<br>5<br>B<br>10.811      | carbon<br>6<br>C<br>12.011     | nitrogen<br>7<br>N<br>14.007    | oxygen<br>8<br>O<br>15.999    | fluorine<br>9<br>F<br>18.998   | neon<br>10<br>Ne<br>20.180        |
|                               |                               | aluminum<br>13<br>Al<br>26.982 | silicon<br>14<br>Si<br>28.086  | phosphorus<br>15<br>P<br>30.974 | sulfur<br>16<br>S<br>32.065   | chlorine<br>17<br>Cl<br>35.453 | argon<br>18<br>Ar<br>39.948       |
| zinc<br>30<br>Zn<br>65.39     | gallium<br>31<br>Ga<br>69.723 | germanium<br>32<br>Ge<br>72.61 | arsenic<br>33<br>As<br>74.922  | selenium<br>34<br>Se<br>78.96   | bromine<br>35<br>Br<br>79.904 | krypton<br>36<br>Kr<br>83.80   |                                   |
| cadmium<br>48<br>Cd<br>112.41 | indium<br>49<br>In<br>114.82  | tin<br>50<br>Sn<br>118.71      | antimony<br>51<br>Sb<br>121.76 | tellurium<br>52<br>Te<br>127.60 | iodine<br>53<br>I<br>126.90   | xenon<br>54<br>Xe<br>131.29    |                                   |

Nobel Gases

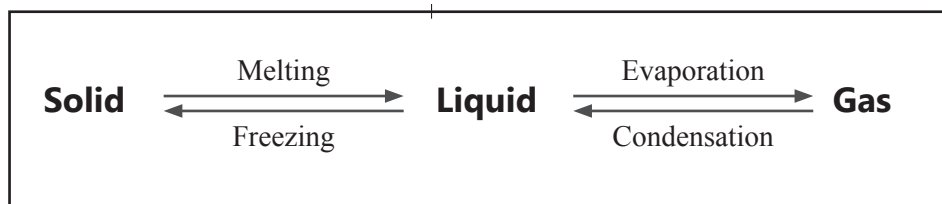
## Phases of Matter

- Atoms** in a **solid** are fixed in a crystalline pattern, **atoms** in a **liquid** can move around each other, and **atoms** in a **gas** exist independently.



Temperature determines phase

Atoms have more kinetic energy as temperature increases



Changing from one state to another is an example of a **Physical Change** (not a chemical change)

### Physical Changes

No new substances created  
e.g. ripping paper  
making a mixture  
melting copper  
changing shape

### Chemical Changes

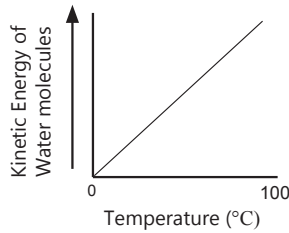
new substances created  
e.g. formation of rust  
burning of fuels  
photosynthesis  
baking cookies

**Mixture**- a combination of substances (e.g. sand and salt in water) that can be separated from each other

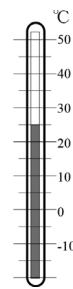
**Solubility**- sugar can dissolve in water. Heating and stirring increase solubility

## Thermal Energy

- Also called heat. As temperature increases, the atoms and molecules move faster and thermal energy increases.



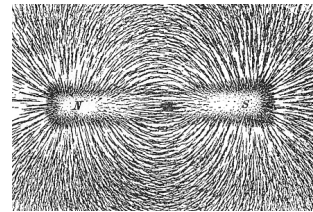
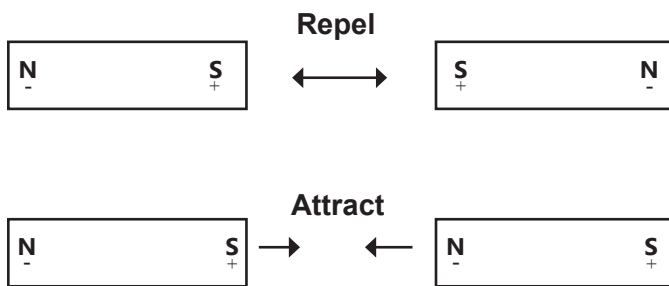
Thermometer



The liquid expands when heated and contracts when cooled.

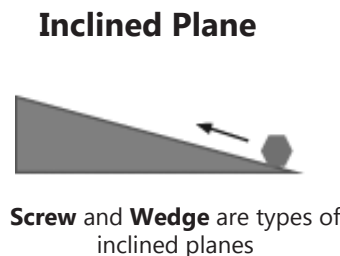
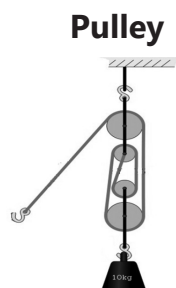
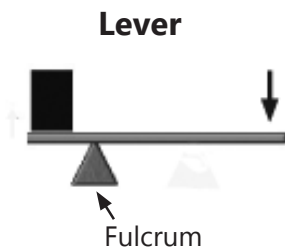
**Convection currents** occur in the mantle of the Earth

## Magnets



The field is strongest at the poles

## Simple Machines

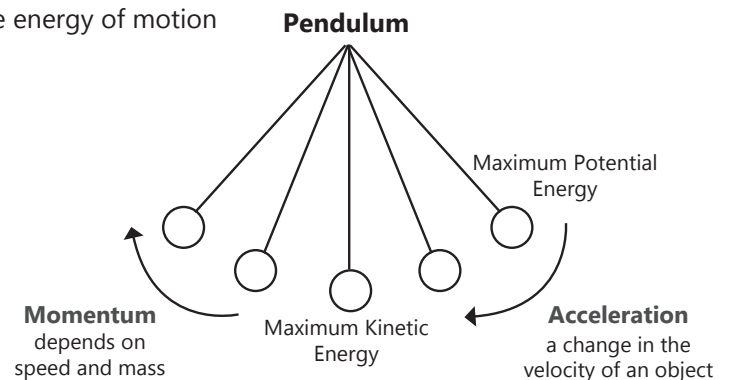
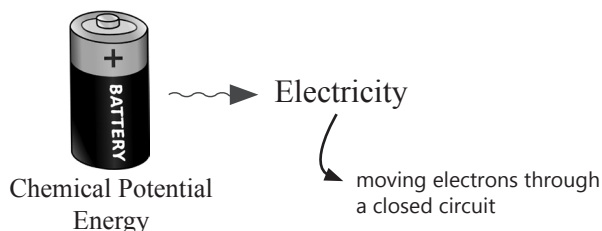


Pulleys and levers can *change the direction* of the force

## Motion and Energy

**Potential Energy** - stored energy    **Kinetic Energy** - the energy of motion

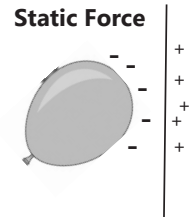
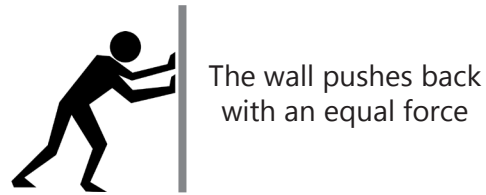
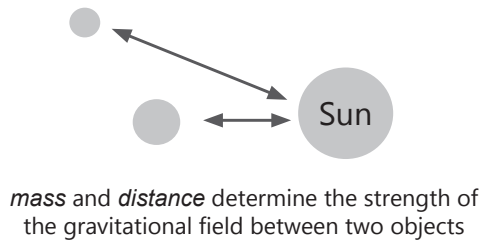
- Machines use **Mechanical Energy**
- Energy can be converted from one form into another (solar energy, photosynthesis)



## Forces

- a push or pull on an object. A force usually changes the motion of an object
- every action has an equal and opposite reaction.

**Gravity**- an attractive force between two objects



**Friction** — a force that opposes motion (air resistance, drag)  
— friction usually produces thermal energy

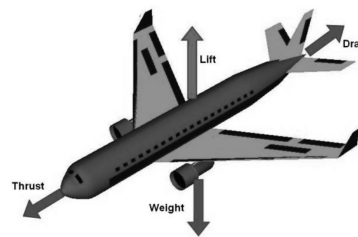
**Lift** an upwards force that affects objects in motion in a fluid

**Thrust** - a force that propels rockets and airplanes

**Drag**- a force that opposes thrust, a type of friction

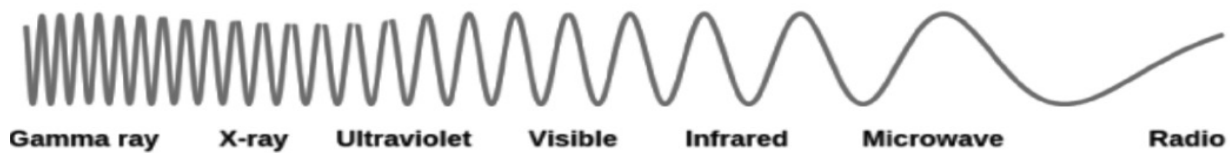
**Weight**- the pull of gravity

### Forces Involved in Flight



**Spring Scale**  
measures  
force

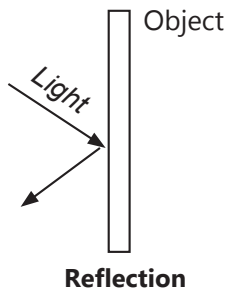
## Electromagnetic Waves



Short Wavelengths —————> Long Wavelengths

- The shorter the wavelength, the more energy carried by the waves
- Visible light is the portion of the EM spectrum that can be detected by the human eye

### Light Waves



### Sound Waves

- Sound waves are produced by **mechanical vibrations**
- Sound waves need a solid, liquid or gas to travel through and cannot travel through space (a vacuum)

**Lightning** - light waves, arrive first

**Thunder** - sound waves, arrive second